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To Theory And
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By Kenneth A
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description of any
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employs the four laws
of thermodynamics
that form an
axiomatic basis the
first law specifies
that energy can be
exchanged between
physical systems as
heat and work the
second law defines
the existence of a
quantity called
entropy that

describes the
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that a system can
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substance 1 3 work
and heat 1 4 the first
law of thermodynamics
1 5 the second law of
thermodynamics 1 6
the steam cycle''**

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in solubility depends
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constants of the
pharmaceutical itself
as well as on the
amount and nature of
the acid or base
added another
possibility is to
form a so called
solid solution in
which the
pharmaceutical pound
is integrated into an
amorphous hydrophilic
polymer which acts as
carrier matrix'

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written in an easily
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boundary of the system can be fixed or it can be movable between the system and surrounding the exchange of mass or energy or both can occur types of thermodynamic systems'

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system in

thermodynamics the system is defined as a definite space or area on which the study of energy transfer and energy conversions is made open system system in which both mass and energy cross the boundaries of the system closed system system in which mass does not cross boundaries of the system though energy may do

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thermodynamics is the study of the relationship between heat or energy and work in other words thermodynamics looks at how we can put energy into a system whether it is a machine or a molecule and make it do work'

'1 1 what it s all about

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thermodynamics is a

science and more

importantly an

engineering tool used

to describe processes

that involve changes

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transfer the unifying
theme is the
application of these
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