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*Name Answer each of the following
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Consider the chemical reaction

represented by the equation below $3 \text{MgCl}_2 + 2 \text{Al} \rightarrow 3 \text{Mg} + 2 \text{AlCl}_3$

3 a If 8 moles of magnesium chloride react with enough aluminum how many moles"

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each word equation uses chemical names What is the mole ratio of MgCl_2

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Key 1 How many moles are in 15 grams of

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